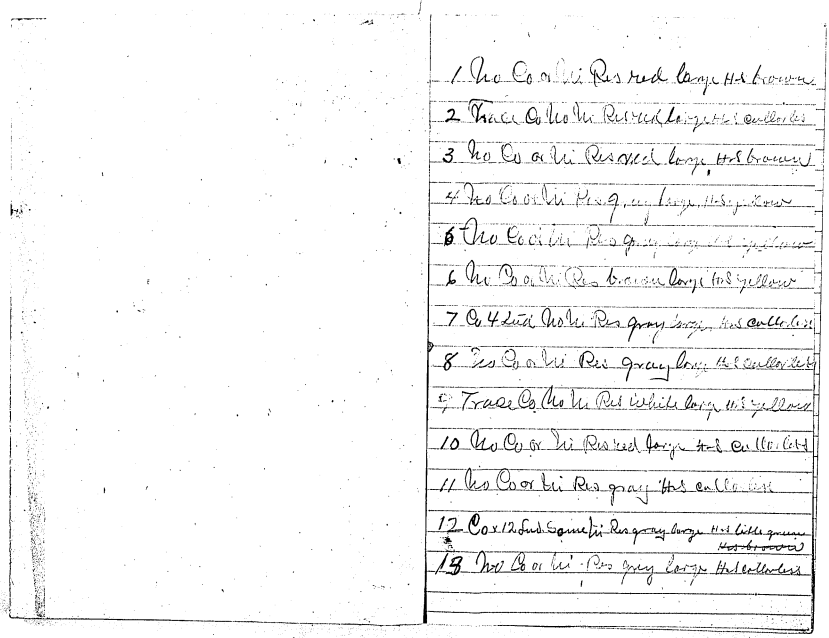
Notebook, N-06-00-00.3


1 No Co or Li Res red large Hs brown
2 Trace Co No Li Res red large Hs Calico Cat
3 No Co or Li Res red large Hs brown
4 No Co or Li Res gray large Hs yellow
5 No Co or Li Res gray large Hs yellow
6 No Co or Li Res brown large Hs yellow
7 Co 4 Lit No Li Res gray large Hs calico cat
8 Tri Co or Li Res gray brown Hs Calico Cat
9 Trace Co No Li Res White large Hs yellow
10 No Co or Li Res red large Hs Calico Cat
11 No Co or Li Res gray Hs calico cat
12 Cox 12 Lit Co No Li Res gray large Hs little gray Hs brown
13 No Co or Li Res gray large Hs calico cat

14

15 Prac Co Ni Li Ro brown large Hi S polish

16 Co x 3% Cu Ni Co Ro grey large Hi S polish

17

18 Co 100 x Cu Ni Co Ni Ro grey large Hi S polish

19 Co 100 x Cu Ni Co Ni Ro grey large Hi S polish

20 Co 0.6% Ni Cu Ro grey large Hi S polish

21 Co 2 Ox Cu Ni Cu Ro black large Hi S little polish

22 Co 0.02% Ni Cu Ro black large Hi S little polish

23 Co 0.1% Ni Cu Ro grey large Hi S little polish

24 Co 0.1% Ni Cu Ro black Hi S polish

25 Co 0.075% Ni Cu Ro black large Hi S polish

26 Ni Co Ox Cu Ro Rock large

27 Co 0.33% No Li Res black large

28 No Co or Li R. brown large Has pinks

29 No Co or Li Res gray large Has orange

30 No Co or Li Res black large Has Collodes

31 No Co or Li Res Gray large Collodes

32 Co 0.01% No Li Res gray large

33 No Co or Li Res Red large Has collodes

34 Trace Co No Li Res gray large Has collodes

35 Co 0.1% No Li Res brown large Has Collodes

36 Trace Co No Li Res yellow large Has Collodes

37 No Co or Li Res gray large Has collodes

38 No Co or Li Res gray large Has green

39 No Co or Li Res brown large Has gray

40 No Co or Li Res gray large Has green

41 No Co or Li Res red large, H2S colorless

42 No Co or Li Res gray large H2S colorless

wood 43 No Co or Li Res brown large H2S colorless

44 No Co or Li Res green large H2S colorless

wood 45 No Co or Li Res gray large H2S colorless

wood 46 Co 0.05% Li Li Res gray large H2S colorless

47 No Co or Li Res green large H2S colorless

48 Trace Co Li Li Res gray large H2S colorless

49 No Co or Li Res gray large H2S colorless

50 Co 1.5% Some Li Res brown large H2S very pink

51 Co 0.05% No Li Res gray large H2S colorless

52 Trace Co No Li Res brown large H2S colorless

53 No Co or Li Res gray large H2S green

54 No Co or Li Res gray large H2S colorless

Specs from No 1 Carbon Glandenning
fined with Reg Copper Wells assay
Crushed 200 mesh roasted Kek
all Sol HCl Conc -

4 Tests.

No 1 50 grms roasted ore 20 cc H_2SO_4 Sp Gr
200 cc water -

No 2 Dup Except 22.5 cc H_2SO_4

No 3 Dup Except 25 cc H_2SO_4

No 4 Dup Except 27.5 cc H_2SO_4

After nearly 6 oc hrs for 6 hours
Results of No 1 weighed 39.750
" 2 " 38.250
" 3 " 37.000
" 4 " 37.000

Evidently this batch of ore did not require
much or SO4 to dissolve -

I put the residues together 154 grammes with 115 cc H_2SO_4 + 500 cc H_2O - *

All these Expts above vitiated by using wrong bottle & getting dilute instead of Conc H_2SO_4 -

* Expmt on dissolving residue how Conc acid -

Fractional precipn -

Na_2CO_3 no difference

Sulphite Na No diffn

Arsenate Na - precips Cobalt + leaves nickel (possibly to make perfect separation -

Phosphate Soda —

Thus acts Curious growing all Kinds of precips — should be investigated —

Tungstate Soda — Curious

Mi alone gives white with ground lungs Liquor Cloar Cobalt alone pink if I remember right no lungs to follow,

But if Cobalt Co + Mi Salfate in Sol, The fullbody is deep green cloar — Thus is with Thiosulfate)

Chromate K. = all precips alike in Colci — Dil SO4 - put in precip dis out comes in a leaves black precip — all purples — what is it — ? ; OGr

Think the Black mass dilute H_2SO_4 is Nickel —
Its Soln HCl dilute & gives green fellowade — The least precip clear HCl & NH_4OH H_2O_2 added gives strong Co ppt solution —

Dried Mixed Sol Co Ni SO_4 — with Boracic acid mixed then precipd in 2 lots by KOH & Cooling 1st precip'n appears Ni has the Ni & 2nd the Co — by physical properties strong Cooling & Re Cooling 1st precip probably Ni Can be got free from Co Theo looks OK —

The best operation (for Comel) of Ni from Co is by crystals of the Ni & Ni ammon sulphate Leave Co in solution + then Recrystals 2 or more Times —

Expmt-

200 grms Can 1 - Dr. treated HCl dec Lbs - roasted residue 325 grms - Used 200 cc 1:1 SG H_2SO_4 Cook 16 hours hot plate (evaporation fallen) - residue dried & roasted so most ~~of~~ SO_3 Come off - not fully as used Evap dish dry flame right on balloons Weight 37.740 grms 162.26 grms dissolved Think if all SO_3 driven off probably residue 62 30 or less grms

Boiled above residue in Water - filtered trace Silver came out some Co & Fe_2 only trace Co recovered

Thus makes	1000	grms Ore
	670	Leaches Lime out
	325	Roasted
	55	Residue

Grow white Residue to wet put in 30 cc Conc HNO_3 + add 10 to 15 cc H_2O & heat - No Cobalt & very little Fe_2 if any Come out - Some Silver shows

1 / 37.750

200 000
37742
162.26

A W Gemmill
The Anglo French Nickel Co Ltd
Hafod Isha Works
Swansea Engl

Samp gets Wand from New Caledonia
100 fr for 4½% oxs (CoO 4 grms 194 mtl)
4 francs increase for each tenth
of a unit up to 5%
over 5% 5 francs per tenth
of a unit.

Rent mm at Noumea
to
Thos Manning
Noumea
N Caledonia

for 4½% if oxide CoO - 28.3 cents Noumea
with Fe 16.2 by 36.8 cents,
40 cents Ni CoO for Nickel up 1½2% = 50 cents Ni Metal above 5% —

After H_2O & HNO_3 Extraction residue
weighs 29.370 grms its whitish

Weight after $CoSO_4$ residue roasted ½
hour ~~dull~~ red heat weighed
270 grammes —

After Extraction with boiling water + abt
10% BC H_2SO_4 ~~— most of the~~
Weight residue 43 grammes,
Showing that 227 grammes went
into solution,

Took 10 grms of the Dried 43 grms +
boiled with water No Co but Came out,
it does — Dilute H_2SO_4 — Lets arsenic
Comes out with H_2S —
Apparently no Ag in it —
Trace of Bismuth, guess Bi along in it,

10 grms of Residue from diss in H_2SO_4
has very little Co + no As_2O_3;
has a Black H_2S precip which
prob Copper —

The 43 gms not water sol
after Ca Br$_2$ leached out
Oblean Co-ammonia Co - 10 gms
gives Evap dish brn free NH$_4$ Sal
deep yellow red -
Roasted thins to yellow heat to cancel
a trace of Arsenic left,
Was black Oxide by H6 -

Thus show Arsenic Fe can be
fully roasted to Fe_2O_3 + As driven
off

Balbach's Terms

Ore running	400 oz Ag per ton	Smelting	$6.00
	400@500	"	5=00
	500@600	"	4 00
	600@700	"	3 =0
	700@800	"	2 =0
	Bottom Rocky		

Settlement to be based on assay of Ledoux & Co Weights & Marks as found on our work in presence of Ledoux & Co receive Cal — to be taken & 95% of the Ag Contents to be paid for at Net Market price for Ag ruling on day of receipt of Ledoux assay Certificate showing our Cost + Marks Settlement 14 days after date of such assay Cert or Cash less 1% — Deliver Fob Newark Works

Balbach Smelting & Refining Co